### Chemistry

College of Science, Engineering and Technology
Department of Chemistry & Geology
242 Trafton Science Center N • 507-389-1963

Chair: Brian L. Groh

Mary Hadley, Michael J. Lusch, Marie K. Pomije, Jeffrey R. Pribyl, Danaé Quirk Dorr, James Rife, Theresa Salerno, Daniel Swart, John B. Thoemke, Trent Vorlicek

The department is recognized by the American Chemical Society and offers a BS major that is approved by that organization. Anyone considering a chemistry or biochemistry major or chemistry minor should choose a departmental faculty member as an advisor and consult that advisor often throughout the course of study.

Admission to Major. Admission to a program is necessary before enrolling in 300- and 400-level courses. Admission is granted by the department. To be eligible for admission to the chemistry program, a student must have declared Chemistry or Chemistry Teaching as a first major; completed 32 credits including CHEM 201 and 202 and achieved a minimum GPA of 2.0. Students should also have an assigned chemistry advisor with whom they have discussed the program. Applications for admission to the chemistry program are available in the department office.

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### CHEMISTRY BA

<table>
<thead>
<tr>
<th>Required General Education (3-4 credits):</th>
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<tbody>
<tr>
<td>MATH 113 Trigonometry (3) or</td>
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<tr>
<td>MATH 115 Precalculus Mathematics (4) or</td>
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<tr>
<td>MATH 121 Calculus I (4)</td>
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<table>
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<tr>
<th>Required Support Courses (4-5 credits):</th>
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<tbody>
<tr>
<td>PHYS 211 Principles of Physics (4) or</td>
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<tr>
<td>PHYS 221 General Physics I (5)</td>
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<table>
<thead>
<tr>
<th>Required for Major (Core, 27 credits):</th>
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<tbody>
<tr>
<td>CHEM 201 General Chemistry I (5)</td>
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<tr>
<td>CHEM 202 General Chemistry II (5)</td>
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<tr>
<td>CHEM 305 Analytical Chemistry (4)</td>
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<tr>
<td>CHEM 320 Organic Chemistry I (with lab) (5)</td>
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<tr>
<td>CHEM 321 Organic Chemistry II (2)</td>
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<tr>
<td>CHEM 331 Organic Chemistry II Lab (1)</td>
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<tr>
<td>CHEM 381 Introduction to Research (2)</td>
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<tr>
<td>CHEM 412 Intermediate Inorganic Chemistry (2)</td>
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<tr>
<td>CHEM 495 Senior Seminar (1)</td>
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<table>
<thead>
<tr>
<th>Required Electives for Major (Chemistry, 6 credits):</th>
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<tbody>
<tr>
<td>Choose a minimum of 6 credits from chemistry or biochemistry courses except CHEM 479 and CHEM 482:</td>
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<tr>
<td>CHEM 300/400 Elective</td>
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<tr>
<td>CHEM 300/400 Elective</td>
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<tr>
<th>Required for Bachelor of Arts (BA) degree ONLY: Language (8)</th>
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<tbody>
<tr>
<td>Required Minor: Yes.</td>
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### CHEMISTRY BS

The Chemistry BS major offers students a choice of two options: Option I and Option II: ACS Approved.

**OPTION I**

Option I is for students who want a rigorous preparation in chemistry, but who do not need as comprehensive a program as that prescribed for the A.C.S. option.

<table>
<thead>
<tr>
<th>Required General Education (9 credits):</th>
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<tbody>
<tr>
<td>MATH 121 Calculus I (4)</td>
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### Required Electives for Major (Chemistry, 4 credits):

<table>
<thead>
<tr>
<th>CHEM 495 Senior Seminar (1)</th>
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<tbody>
<tr>
<td>CHEM 451 Physical Chemistry Laboratory II (1)</td>
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<tr>
<td>CHEM 450 Physical Chemistry Laboratory I (1)</td>
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<tr>
<td>CHEM 495 Senior Seminar (1)</td>
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</tbody>
</table>

### Required Electives (Physics or Mathematics, 3-4 credits):

<table>
<thead>
<tr>
<th>PHYS 221 General Physics I (5)</th>
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<tr>
<td>PHYS 222 General Physics II (5)</td>
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</table>

### Required for Major (Core, 50-51 credits):

<table>
<thead>
<tr>
<th>PHYS 221 General Physics I (5)</th>
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<tbody>
<tr>
<td>PHYS 222 General Physics II (5)</td>
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</table>

### Required Support Courses (9 credits): MATH 122 Calculus II (4)

### Required for Major (Core, 41 credits):

<table>
<thead>
<tr>
<th>PHYS 221 General Physics I (5)</th>
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<tbody>
<tr>
<td>PHYS 222 General Physics II (5)</td>
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</table>

### Required for Major (Core, 50-51 credits): MATH 122 Calculus II (4)

### Required Support Courses (9 credits): MATH 121 Calculus I (4)

### Required for Major (Core, 41 credits): MATH 122 Calculus II (4)

### Required for Major (Core, 50-51 credits): MATH 121 Calculus I (4)

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**OPTION II: A.C.S. APPROVED**

The BS Chemistry, A.C.S. option approved by the American Chemical Society is intended for professional chemists and provides an excellent preparation for graduate or professional school, industry or business. Any deviations from this program requires prior approval from the department.

<table>
<thead>
<tr>
<th>Required General Education (9 credits):</th>
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<tbody>
<tr>
<td>MATH 121 Calculus I (4)</td>
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</table>

### Required for Major (Core, 50-51 credits): MATH 122 Calculus II (4)

### Required Support Courses (9 credits): MATH 121 Calculus I (4)

### Required for Major (Core, 50-51 credits): MATH 122 Calculus II (4)

### Required for Major (Core, 50-51 credits): MATH 121 Calculus I (4)

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### Required for Major (Chemistry, 1 credits):

<table>
<thead>
<tr>
<th>CHEM 460 Biochemistry I (3)</th>
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<tr>
<td>CHEM 475 Instrumental Analysis (4)</td>
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<tr>
<td>CHEM 495 Senior Seminar (1)</td>
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</table>

### Required Electives for Major (Chemistry, 1 credits):

<table>
<thead>
<tr>
<th>CHEM 460 Biochemistry I (3)</th>
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</thead>
<tbody>
<tr>
<td>CHEM 475 Instrumental Analysis (4)</td>
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<tr>
<td>CHEM 495 Senior Seminar (1)</td>
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</table>

### Required Electives (Physics or Mathematics, 3-4 credits): MATH 121 Calculus I (4)

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2005-2006 Undergraduate Bulletin
Choose a minimum of 3 credits from the following courses:

<table>
<thead>
<tr>
<th>Course Code</th>
<th>Course Title</th>
<th>Credits</th>
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<tbody>
<tr>
<td>PHYS 441</td>
<td>PHYS 447</td>
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<tr>
<td>PHYS 453</td>
<td>PHYS 473</td>
<td></td>
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<tr>
<td>MATH 321</td>
<td>MATH 455</td>
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</table>

Required Minor: None.

CHEMISTRY MINOR

Required for Minor (Core, 19 credits):

- CHEM 201 General Chemistry I (5)
- CHEM 202 General Chemistry II (5)
- CHEM 305 Analytical Chemistry (4)
- CHEM 320 Organic Chemistry I (with lab) (5)

CHEMISTRY TEACHING BS

Requirements for the Chemistry Teaching BS can be found in the SCIENCE TEACHING section of the bulletin. For information, consult the chemistry education advisor, Jeffrey Pribyl.

POLICIES/INFORMATION

GPA Policy: Students obtaining a major or minor in chemistry must maintain an overall GPA of 2.0 with no more than 5 credits of D work in chemistry courses.

P/N Grading Policy: Courses leading to a major or minor in chemistry or biochemistry may not be taken on a P/N basis except where P/N grading is mandatory.

The first year of coursework for all chemistry and biochemistry majors should include two semesters of chemistry (201, 202) and two semesters of mathematics (selection of courses depends on mathematics background). During the second year, the recommended courses include organic chemistry, advanced mathematics, physics, analytical chemistry. It is important for BS chemistry majors that the calculus and physics sequences be completed by the end of the second year since they are prerequisites for physical chemistry. Physical chemistry and instrumental analysis should be taken during the third year. The advanced courses in chemistry and biochemistry can be taken in the junior and senior years. Participation in chemistry seminar is required of all majors. The coursework in mathematics and physics that is required for a major may be credited toward a major or minors in these areas. For this reason it is often desirable and convenient to choose a joint major or minor with physics or mathematics.

COURSE DESCRIPTIONS

CHEM 100 (4) Chemistry in Society
This course explores and evaluates energy sources from a chemical perspective. This course will examine two of the most significant environmental challenges facing modern society: stratospheric ozone depletion and global climate change, from an interdisciplinary perspective. The course will start by examining, with a minimum of mathematics, the scientific basis and evidence for these phenomena, and then go on to consider the potential implications of and solutions to these challenges. In order to understand these potential implications and solutions, we must realize and understand the interdisciplinary nature of these challenges. Variable GE-3, 9

CHEM 132 (3) Chemistry of Energy
This course explores and evaluates energy sources from a chemical perspective. This chemistry course explores the scientific basis of crime-fighting using physical evidence. Course topics will include discussions of different kinds of evidence, how evidence must be preserved in order to be of value, how to select and analyze samples, and especially how to interpret results of scientific tests. Case studies will be used as examples throughout the course. There will also be discussions of ethical questions about the collection, analysis, and uses of forensic data. Variable GE-3, 9

CHEM 133 (3) Challenges to Our Global Environment
This course explores and evaluates energy sources from a chemical perspective. This course explores and evaluates energy sources from a chemical perspective. This course explores and evaluates energy sources from a chemical perspective. This course will examine two of the most significant environmental challenges facing modern society: stratospheric ozone depletion and global climate change, from an interdisciplinary perspective. The course will start by examining, with a minimum of mathematics, the scientific basis and evidence for these phenomena, and then go on to consider the potential implications of and solutions to these challenges. In order to understand these potential implications and solutions, we must realize and understand the interdisciplinary nature of these challenges. Variable GE-3, 9

CHEM 134 (3) Mind Altering Substances
This course explores and evaluates energy sources from a chemical perspective. This course explores and evaluates energy sources from a chemical perspective. This course explores and evaluates energy sources from a chemical perspective. This course will explore the scientific, pharmacological, neurochemical and cultural aspects of psychoactive substances. The material is presented intuitively, with no mathematics. Course topics will include discussions of the major classes of pharmaceutical and psychoactive substances, basic neurochemistry, the role of psychoactive substances in medicine, the ritual use of psychoactive substances by traditional cultures, the FDA approval process, the significance and implications of drug testing, the controversy of drug-induced behavioral modification, national and global perspectives of substance abuse and the ethics of legalization. Variable GE-3

CHEM 131 (3) Forensic Science
This course explores and evaluates energy sources from a chemical perspective. This course explores and evaluates energy sources from a chemical perspective. This course explores and evaluates energy sources from a chemical perspective. This course will explore the scientific, pharmacological, neurochemical and cultural aspects of psychoactive substances. The material is presented intuitively, with no mathematics. Course topics will include discussions of the major classes of pharmaceutical and psychoactive substances, basic neurochemistry, the role of psychoactive substances in medicine, the ritual use of psychoactive substances by traditional cultures, the FDA approval process, the significance and implications of drug testing, the controversy of drug-induced behavioral modification, national and global perspectives of substance abuse and the ethics of legalization. Variable GE-3

CHEM 201 (5) General Chemistry I
Introduction to the basic principles of chemistry including atomic and molecular structure, bonding, chemical reactions, stoichiometry, thermodynamics and states of matter. Laboratory will reinforce lecture concepts. Pre: MATH 112 or equivalent; high school Chemistry, CHEM 105 or instructor’s consent F,S GE-2, 3

CHEM 202 (5) General Chemistry II
Continuation of the basic principles of chemistry including properties of solutions, kinetics, acids and bases, equilibria, buffers, precipitation reactions, electron transfer reactions, electrochemistry, entropy and free energy. Laboratory will reinforce lecture concepts. Pre: CHEM 201 F,S

CHEM 299 (1-6) Individual Study

CHEM 305 (4) Analytical Chemistry
Introduction to the principles of chemical analysis, with emphasis on classical methods of analysis. Lectures will stress the theory of chemical measurements and sample handling. Laboratory exercises will provide students with opportunities to explore calibration methods, method development, and established procedures for volumetric and gravimetric analyses. Basic atomic spectroscopy is also presented. Pre: CHEM 202 F,S
CHEM 320 (5) Organic Chemistry I
Introduction to organic structure, bonding, chemical reactivity, reactions as acids and bases, mechanisms and stereochemistry. The chemistry of alkanes, alkyl halides, alkenes, alkynes, alcohols, ethers, aldehydes and ketones, carboxylic acids and their derivatives, and amines will be covered. Laboratory illustrates synthetic techniques and the preparation and reactions of functional groups discussed during lecture.
Pre: CHEM 202 F, S

CHEM 321 (2) Organic Chemistry II
The chemistry of aromatic compounds, free radicals, polyenes, macromolecules, heterocyclic compounds, carbohydrates, amino acids, peptides, and proteins will be covered. This will include a study of mechanisms, synthetic transformations, concerted reactions, and spectroscopy.
Pre: CHEM 320 S

CHEM 331 (1) Organic Chemistry II Lab
Laboratory illustrating electrophilic aromatic substitutions and other reactions of aromatic compounds, synthetic transformations as well as qualitative organic analysis.
Pre: CHEM 321 previously or concurrently S

CHEM 360 (4) Principles of Biochemistry
Analysis of the structure and metabolism of biologically important compounds. This intermediate-level course is designed for students in the medical technology, food science, chemistry education, chemistry and pre-professional health majors. The laboratory teaches basic biochemical techniques.
Pre: CHEM 320 S

CHEM 381 (2) Introduction to Research
Introduction to the use of chemical literature (in print and electronic media), current departmental faculty research interests, safe and ethical conduct of laboratory research, and proper recording of research results in laboratory notebooks. Students perform a literature search and write a proposal for an undergraduate research project.
Pre: CHEM 321 F

CHEM 407 (3) Water Chemistry
A broad introduction to the chemistry of natural waters and chemical analysis of such systems. Topics covered may include: macromolecular analytes, organic analytes, inorganic analytes, major component/minor component/trace component determinations, matrix effects, equilibrium processes, modeling of chemical/physical transport, regulatory monitoring, and compliance issues. Laboratory exercises will provide students with goal-oriented, cooperative experiences in sampling and measurement of complex samples.
Pre: CHEM 305 Variable

CHEM 412 (2) Intermediate Inorganic Chemistry
Use of the principles of chemistry such as atomic structure, bonding, thermodynamics and acid-base behavior to focus on the properties of the more interesting, important and unusual elements and compounds. Emphasis will be placed on the representative elements and selected transition elements.
Pre: CHEM 320 S

CHEM 413 (3) Advanced Inorganic Chemistry
A survey of topics in inorganic chemistry including quantum mechanics, symmetry and group theory, solid state chemistry, molecular structure and geometry, bonding theories, and coordination chemistry, emphasizing the theoretical foundation.
Pre: CHEM 440 F

CHEM 415 (2) Inorganic Preparations
The preparation and study of inorganic/organometallic compounds utilizing a variety of synthetic techniques including common Schlenk techniques. The studies will include characterization by common instrumental methods such as IR, NMR, and UV-vis spectroscopy. Additional studies using instrumental techniques such as IR, NMR, UV-vis, electrochemistry and magnetic susceptibility will also be conducted.
Pre: CHEM 413 S

CHEM 423 (4) Chemical and Spectroscopic Determination of Structure
Spectroscopic techniques including nuclear magnetic resonance, infrared, and mass spectrometry for determining structural features of molecules will be covered. Spectroscopic methods emphasize interpretation of spectra, and also provide hands-on operation of the corresponding electronic instruments. The laboratory uses these techniques for the determination of the structures of a series of unknown compounds.
Pre: CHEM 321 and 331 F

CHEM 424 (3) Advanced Organic Chemistry
Advanced synthetic organic reactions and their mechanisms. Laboratory will include examples of some of this chemistry, and techniques for reaction monitoring and product purification.
Pre: CHEM 423 S-EVEN

CHEM 432 (2) Industrial Chemistry
The synthesis and properties of organic macromolecules, especially industrially important polymers, and the chemistry of other industrially important chemical reactions and processes.
Pre: CHEM 321 S-ODD

CHEM 437 (4) Food Chemistry
This lecture laboratory course will cover the fundamental principles of food chemistry. Chemical and physical properties of major and minor food components will be discussed. The laboratory will involve both traditional wet chemical methods and more sophisticated instrumental analyses.
Pre: CHEM 305, 320; Pre or Co: CHEM 360 or 460

CHEM 440 (3) Physical Chemistry I
Detailed treatment of thermodynamics and chemical kinetics. Topics include equations of state, laws of thermodynamics, statistical thermodynamics, phase and reaction equilibrium, thermodynamics of solutions and electrochemistry, transport properties, and reaction kinetics.
Pre: CHEM 305, 321, one year of physics, MATH 121 F

CHEM 441 (3) Physical Chemistry II
Detailed treatment of quantum mechanics, spectroscopy, and statistical mechanics. Topics include the foundations of quantum mechanics, application of quantum mechanics to atomic and molecular structure, foundations of spectroscopic techniques and statistical mechanics.
Pre: CHEM 440, MATH 122 S

CHEM 445 (2) Advanced Physical Chemistry
Integrated application of the content from 440 and 441 to an applied topic of interest to the instructor. The course will depend heavily on reading and discussion of current primary literature of physical chemistry. Possible topics include: atmospheric chemistry, thermodynamics of protein folding, catalytic processes, or molecular processes at interfaces.
Pre: CHEM 441 Variable

CHEM 450 (1) Physical Chemistry Laboratory I
Laboratory to accompany 440. An advanced treatment of measurement theory and data analysis precedes a series of thermodynamic and kinetic experiments designed to complement topics treated in lecture to help students’ independence and sophistication in planning, performing, and reporting experimental work.
Pre: CHEM 440 previously or concurrently F

CHEM 451 (1) Physical Chemistry Laboratory II
Laboratory to accompany 441. Experiments and computational projects in quantum mechanics, spectroscopy, and statistical mechanics. The experiments and projects will continue to work toward the goal of increasing the students’ independence and sophistication.
Pre: CHEM 441 previously or concurrently S

CHEM 460 (3) Biochemistry I
Detailed analysis of the structures, properties, and functions of proteins, carbohydrates, and lipids; introduction to carbohydrate metabolism; theory for the purification and analysis of proteins. Concurrent enrollment in CHEM 465 is recommended.
Pre: CHEM 320, and BIOL 106 F

CHEM 461 (3) Biochemistry II
Detailed analysis of the reactions involved in intermediary metabolism, translation, transcription, and replication.
Pre: CHEM 460 S

CHEM 465 (1) Biochemical Techniques I
A lecture/laboratory course which presents methodology and instrumentation used to purify and analyze biomolecules. Techniques include chromatography, autoradiography, and radioisotope techniques, polyacrylamide gel electrophoresis, and spectrophotometry.
Pre: CHEM 460 previously or concurrently. CHEM 305 is recommended. F

CHEM 466 (2) Biochemical Techniques II
Students work in teams to solve biochemical research problems by analyzing data from experiments which they design.
Pre: CHEM 460 and 465 S

CHEM 474 (2) Chromatography
Theory and applications of thin layer, paper, liquid, gas and supercritical fluid chromatography and capillary electrophoresis.
Pre: CHEM 320 previously or concurrently is recommended F-EVEN

CHEM 475 (4) Instrumental Analysis
Theory and practice of modern instrumental methods including basic electronics. Special emphasis placed on sampling methods, analog and digital electronics, electrochemistry, spectrophotometric, and chromatographic methods, surface and thin-film analysis and computer acquisition and data processing techniques.
Pre: CHEM 305; PHYS 212 or 222 is recommended S

CHEM 477 (1-3) Special Topics in Instrumental Analytical Chemistry
Detailed study and focused discussion of a specific analytical technique such as electrochemistry, X-ray analysis, etc. or an area of analysis such as metals, bioanalytical, etc. May be taken more than once for credit.
Pre: CHEM 305 Variable

CHEM 479 (4) Teaching Physical Science
Methods and materials for teaching physical sciences in middle school through high school. Clinical experiences required for the course.
Pre: Consent S

CHEM 482 (1-3) Problems in Teaching Science
Variable

CHEM 485 (1-2) Seminar in Environmental Chemistry
Study of current environmental problems or issues with emphasis on the relevant chemical needs and understanding necessary to monitoring or alleviating the problems.
Pre: CHEM 305 Variable

CHEM 490 (1-6) Workshop

CHEM 495 (1) Senior Seminar
Capstone course for majors in Chemistry, Biochemistry, and Chemistry Teaching. During this course students will present the results of their research in several different forums including oral presentations and poster sessions.
Pre: Consent S

CHEM 496 (1-6) Senior Thesis

CHEM 497 (1-16) Internship

CHEM 498 (1-6) Undergraduate Research

CHEM 499 (1-6) Individual Study